

Hydrogen Peroxide

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Researchers have developed an **efficient**, **eco-friendly**, **and energy-saving** method for synthesizing **hydrogen peroxide** (H₂O₂). using a **hydrazone-linked Covalent Organic Framework** (COF).

- The COF enables H₂O₂ production under visible light without external sacrificial electron donors or additional reagents, making it a cleaner and more efficient alternative to conventional methods.
- About Hydrogen Peroxide (H₂O₂): It is a colorless liquid with a bitter taste at room temperature, highly unstable, and decomposes into oxygen and water, releasing heat.
- Uses: Antiseptic for wounds (diluted solutions), disinfectant in hospitals and clinics.
 - Industrial Uses: Bleaching agent in paper, textile, and cosmetic industries.
 - Rocket Propulsion: High-concentration H₂O₂ used as a propellant.
 - Chemical Applications: Employed in food processing for sterilization.
 - Hazards: It is a strong oxidizer and may cause spontaneous ignition with combustible materials.
- Peroxide chemicals are organic compounds that contain a peroxide functional group (two linked oxygen atoms).

Read More: Explosive Substances Act and Peroxide Chemicals

PDF Reference URL: https://www.drishtiias.com/printpdf/hydrogen-peroxide